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completely reacts with tin, as shown in the following reaction equation: $\text{Sn}(s) +$



_____ The production of ammonia is given by the equation: $3\text{H}_2(g) + \text{N}_2(g)$

$2\text{NH}_3(g)$ If 5.0 g of hydrogen reacts with excess nitrogen, how much ammonia will be produced?

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